

Anhydrous 1-methylimidazolidin-2-one

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Key indicators

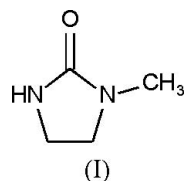
Single-crystal X-ray study
 $T = 298$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.070
 wR factor = 0.161
Data-to-parameter ratio = 13.7For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

The title compound, $\text{C}_4\text{H}_8\text{N}_2\text{O}$, crystallizes from the vapor phase in sheets supported by a one-dimensional hydrogen-bonded network. The molecular unit shows planar ureido $-\text{NCH}_3$ groups, but the slight out-of-plane hydrogen-bond geometry may be indicative of some pyramidalization of the ureido $-\text{NH}$ groups.

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Comment

The imidazolidin-2-one functional group is of primary importance in the transfer of carbon dioxide by biotin-dependent enzymes (Carey *et al.*, 2004; Attwood & Wallace, 2002). It is therefore important that the structural characteristics of this unit be well understood. In the course of our work on biomimetic carbon dioxide fixation, we crystallized anhydrous 1-methylimidazolidin-2-one, (I), and report here its crystal structure.



The molecular unit in (I) shows the five-membered heterocycle. The ring is only slightly enveloped, with an average internal torsion angle of 4.6° and a mean deviation from planarity of 0.0273 Å. The metrical parameters in the heterocycle are essentially identical to those in unsubstituted imidazolidin-2-one (Kapon & Reisner, 1989), and to those in biotin itself (DeTitta *et al.*, 1976). Methylated atom N2 is planar, indicative of largely sp^2 -hybrid character. This is consistent with N -silyl-substituted imidazolidin-2-ones (Szalay *et al.*, 2005), as well as with $N1'$ -methoxycarbonylbiotin methyl

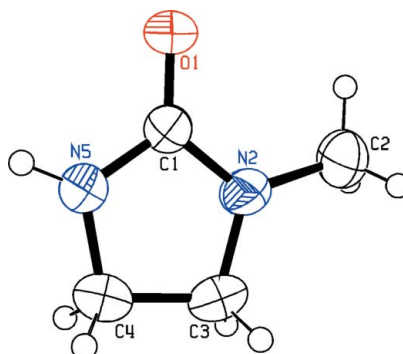


Figure 1
Displacement ellipsoid drawing of (I) (35% probability ellipsoids).

ester (Stallings *et al.*, 1980), which also exhibit essentially planar ureido N atoms. However, the planar N atoms are at variance with some theoretical work which seems to indicate that the N atoms in urea and related molecules have considerable sp^3 character (Meier & Coussens, 1992). Some insight may be gained from the solid state packing in (I), which is supported by a one-dimensional network of hydrogen bonds ($O \cdots N = 2.87 \text{ \AA}$). Neighboring hydrogen-bonded chains are packed to form a two-dimensional sheet (Fig. 2). The individual hydrogen-bonding interactions are inclined at 19.8° to the heterocycle plane, and may be indicative of some pyramidalization of the $-\text{NH}$ group.

Experimental

The title compound was purchased from Aldrich. The microcrystalline powder was placed in a flame-sealed ampoule and the apparatus stored in a 363 K oven for one week. This procedure gave colorless blocks suitable for X-ray diffraction.

Crystal data

$\text{C}_4\text{H}_8\text{N}_2\text{O}$	$D_x = 1.321 \text{ Mg m}^{-3}$
$M_r = 100.12$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 1593 reflections
$a = 7.5746 (11) \text{ \AA}$	$\theta = 3.0\text{--}31.6^\circ$
$b = 9.3984 (13) \text{ \AA}$	$\mu = 0.10 \text{ mm}^{-1}$
$c = 7.9482 (11) \text{ \AA}$	$T = 298 (2) \text{ K}$
$\beta = 117.139 (2)^\circ$	Block, colorless
$V = 503.53 (12) \text{ \AA}^3$	$0.38 \times 0.30 \times 0.10 \text{ mm}$
$Z = 4$	

Data collection

Bruker SMART APEX diffractometer	888 independent reflections
ω scans	712 reflections with $I > 2\sigma(I)$
Absorption correction: ψ scan (Blessing, 1995)	$R_{\text{int}} = 0.092$
$T_{\text{min}} = 0.94$, $T_{\text{max}} = 0.99$	$\theta_{\text{max}} = 25.0^\circ$
3853 measured reflections	$h = -9 \rightarrow 9$
	$k = -11 \rightarrow 11$
	$l = -9 \rightarrow 9$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0535P)^2 + 0.1874P]$
$R[F^2 > 2\sigma(F^2)] = 0.070$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.161$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.16$	$\Delta\rho_{\text{max}} = 0.14 \text{ e \AA}^{-3}$
888 reflections	$\Delta\rho_{\text{min}} = -0.18 \text{ e \AA}^{-3}$
65 parameters	
H-atom parameters constrained	

H atoms were placed in idealized positions as riding atoms with C—H distances of 0.96 \AA for methyl and 0.96 \AA for the rest; The N—H distance is 0.86 \AA . The isotropic displacement parameters were set at $1.5U_{\text{eq}}$ of the parent atom for the methyl H atoms and $1.2U_{\text{eq}}$ for the rest.

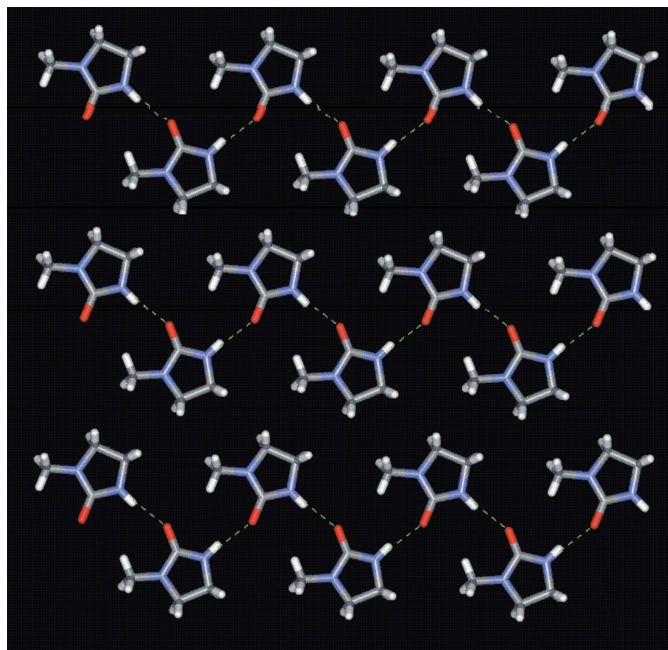


Figure 2

Solid-state packing diagram of (I), viewed down the crystallographic a axis.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT-Plus* (Bruker, 1997); data reduction: *SAINT-Plus*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker 1997); software used to prepare material for publication: *SHELXTL*.

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